## Regiocontrolled Synthesis of Spiro Furopyrone-Cyclohexane Ring Systems : Approach to the Pyranoditerpene Lygodinolide

## Asish K. Banerjee and Basudeb Achari

Indian Institute of Chemical Biology, Calcutta - 700032, India.

Abstract: The isomeric  $\alpha$ - and  $\gamma$ -pyrones 8 and 9 have been synthesised through oxidative cyclisation of the intermediate 5, prepared via Claisen condensation of a protected  $\beta$ -ketoester with lithio ethyl acetate.

Lygodinolide (1), a novel pyranoditerpene, was recently isolated  $^1$  by us as a minor constituent of the fern <u>Lygodium flexuosum</u>, reported  $^2$  to exhibit antifertility activity. The unique structural features and the potential bioactivity of the molecule prompted us to undertake a total synthesis of 1. It was perceived that the construction of the crucial spiro furopyrone ring system with an adjacent methylene substituent would form an ideal initial target. Herein we report the highly regional receive synthesis of the model compound 8 and the isomer 9, containing the spiro furo- $\alpha$ -pyrone and - $\gamma$ -pyrone moieties respectively, through a common intermediate 5.

As outlined in Scheme 1, the synthesis involved Claisen condensation of the ketal ester 2 with the lithic salt of ethyl acetate followed by alkylation of the product  $3^3$  with  $\beta$ -cyclogeranyl bromide to afford 4 (69%) as a diasterecisometric mixture

Reagents and conditions: (i) LiHMDS (2 eqv), ethyl acetate (2 eqv), -78° to 0°C, 2h; (ii) NaH (1.25 eqv),  $\beta$ -cyclogeranyl bromide (1.2 eqv), THF, 0° to 25°C, 16h, then reflux, 3h; (iii) Pd(II) chloride-acetonitrile (2 mole %), acetone, rt, 24h; (iv) DBU (1 eqv), benzene, reflux, 14h; (v) m-CPBA (50%, 2.5 eqv), CaO (6.5 eqv), anh. sodium sulfate (24.3 eqv), dichloromethane, 15° to 25°C, 14h; (vi) m-CPBA (50%, 1.3 eqv), CSA (Cat), dichloromethane, 0° to 25°C, 17h; (vii) sulfury chloride (2 eqv), DBU (5 eqv), dichloromethane, -20°C, 4h.

(although separated chromatographically for characterisation, the mixture as such was used for the subsequent steps). Diketalisation of 4 with  $PdCl_2$  ( $CH_3CN)_2$  in acctone to the desired diketoester and subsequent cyclisation with DBU in benzene then produced the pyrone 5 in 85% yield. This approach for pyrone synthesis permits a wide flexibility in introducing the substituents.

Oxidative cyclisation of 5 with m-chloroperbenzoic acid gave a 3:7 mixture of the isomeric spiro furo- $\alpha$ -pyrone 6 and - $\gamma$ -pyrone 7 separated by column chromatography. The desired selectivity in product formation could however be realised through a remarkable control of reaction conditions. Thus, using a catalytic amount of camphor sulphonic acid, the  $\gamma$ -pyrone was formed almost exclusively (85%). On the other hand, the product ratio could be dramatically altered in favour of the desired isomer 6 (74%) by carrying out the reaction in presence of an acid scavenger (CaO).

Finally, dehydration of 6 led to the olefin 8 (82%) representing the crucial spiro furo-0-pyrone-cyclohexane ring structure of 1, while 7 yielded 9 (85%).

## REFERENCES AND NOTES

- Achari, B.; Chaudhuri, C.; Saha, C.R.; Pakrashi, S.C.; McPhail, D.R.; McPhail, A.T. J. Org. Chem. 1990, 55, 4977-4978.
- 2. Gaitonde, B.B.; Mahajan, R.T. Indian J. Med. Res., 1980, 72, 597-604.
- 3. Selected data for 3-9. UV spectra recorded in MeOH, NMR (100 MHz) in CDCl $_3$ .
  - 3, bp. 160-165°C/10 mm; IR (neat): 1740, 1711 cm<sup>-1</sup>.
  - 4, oil; NMR :  $\delta_{H}$  0.95,0.98,1.31(3x3Hs), 1.12(3Hd,J 7Hz), 1.24(3Ht,J 7Hz), 1.92 (2Hm), 2.65(2Hd,J 7Hz), 3.08(1Hg,J 7Hz), 2.92(5Hm), 4.15(2Hg,J 7Hz).
  - 5, m.p.  $118-119^{\circ}C$ ; UV :  $\lambda_{\text{max}}$  292 nm; IR (KBr) : 3254,  $1671 \text{ cm}^{-1}$ ; NMR :  $\delta_{\text{H}}$  1.02,1.71,1.84,2.20(5x3Hs), 2.04(2Hbrd), 3.28(2Hs), 8.36(1Hs).
  - 6, m.p. 198-200°C; UV : $\lambda_{max}$  297 nm; IR (KBr) : 3398, 1700, 1647 cm<sup>-1</sup>; NMR : $\delta_{H}$  0.78,1.06,1.14,1.90,2.21(5x3Hg), 1.38-1.84(6Hm), 2.92,3.13(2x1Hd,J 16Hz).
  - 7, m.p.  $184-185^{\circ}$ C; UV :  $\lambda_{max}$  265,274 nm: IR (KBr) : 3390,1666,1612 cm<sup>-1</sup>; NMR :  $\delta_{H}$  0.84,1.12,1.16,1.91,2.24(5x3Hg), 2.94,3.20,(2x1Hd,J 15Hz).
  - 8, oil; IR (neet): 1736,1715,1653,869 cm<sup>-1</sup>; NMR:  $\delta_H$  0.88,1.00,1.98,2.22(4x3Hg); 2.87,3.19(2x1Hd,J 15Hz), 4.84(2H<u>brs</u>);  $\delta_C$  170.5,161.9,160.1,146.9,102.8,99.5,98.5,39.0 (8xg), 107.4,36.8,32.4,32.1,21.9(5xt), 22.9,21.7,16.9,9.1(4xg).
  - 9, m.p. 105°C; IR (KBr) : 1676,1630,1602,925 cm<sup>-1</sup>; NMR :  $\delta_{\rm H}$  0.91,1.00,1.92, 2.26 (4x3Hs), 2.90,3.20(2x1Hd,J 15Hz), 4.86,4.96(2x1Hbrs);  $\delta_{\rm C}$  177.0,166.5,154.7,146.1, 119.1,97.7,95.3,38.9(8xs), 107.8,36.4,32.2,30.7,21.7(5xt), 22.5,21.7,16.8,9.7(4xg).
- 4. Lipshutz, B.H.; Pollart, D.; Monforte, J.; Kotsuki, H. <u>Tetrahedron Lett.</u> 1985, 26, 705-708.
- Cervello, J.; Marquiet, J.; Moreno-Manas, M. J. Chem. Soc. Chem. Commun. 1987, 644-645.

## (Received in UK 9 November 1992)